Hryn et al. (S.N. 10/702,419) Response to March 1, 2007 Office Action Page -2-

Amendments to the Claims:

This listing of claims will replace all prior versions, and listings of claims in the application.

- 1. (Previously Presented) An electrolyte for the electrolysis of alumina, the electrolyte comprising a mixture of aluminum fluoride, potassium fluoride, oxygen-containing ions, and NaF, wherein the NaF comprises no more than 2 weight percent of the electrolyte.
- 2. (Previously Presented) The electrolyte recited in claim 1 wherein the potassium fluoride to aluminum fluoride molar ratio ranges from about 1.0 to 1.5.
- 3. (Currently Amended) The electrolyte recited in claim 1, the electrolyte further comprising from about 4 to 6 wt. % of <u>dissolved</u> aluminum oxide (alumina/Al₂O₃).
 - 4. (Canceled)
 - 5. (Canceled)
- 6. (Currently Amended) The electrolyte as recited in claim 3 wherein the electrolyte is a liquid between 600 and 1000 °C.

Claims 7-20 (Canceled)

21. (Currently Amended) A liquid phase electrolyte at between 600 and 1000 °C, the electrolyte comprising potassium fluoride, aluminum fluoride, and less than two weight percent of NaF.

Hryn et al. (S.N. 10/702,419) Response to March 1, 2007 Office Action Page -3-

- 22. (Currently Amended) The electrolyte as recited in claim 21 further comprising from about 4 to 6 wt. % of dissolved aluminum oxide.
- 23. (Currently Amended) The electrolyte as recited in claim 22 further comprising $Al_2OF_6^2$ and $Al_2O_2F_4^{-2}$.
- 24. (Currently Amended) An electrolyte consisting of A liquid electrolyte comprising potassium fluoride and aluminum fluoride in a 1.3 molar weight ratio, between about 4 to 6 wt% dissolved aluminum oxide, and less than two weight percent NaF.
 - 25. (New) A liquid electrolyte produced by the process of
 - a.) supplying a mixture of potassium fluoride, aluminum fluoride, 4-6 weight percent alumina and no more than 2 weight percent NaF; and
 - b.) heating the mixture to between 660 and 1000 °C.
- 26. (New) The liquid electrolyte as recited in claim 25 wherein the molar ratio of potassium fluoride to aluminum fluoride ranges from about 1.0 to 1.5.
 - 27. (New) The liquid electrolyte as recited in claim 25 wherein the alumina is dissolved.